



Tetrahedron 59 (2003) 1791-1796

TETRAHEDRON

Photo-Fries and Fries reaction of 5,8-dihydro-1-naphthyl esters

Kamaraj Sriraghavan and Vayalakkavoor T. Ramakrishnan*

Department of Organic Chemistry, University of Madras, Guindy Campus, Chennai 600 025, India

Received 15 December 2002; revised 16 December 2002; accepted 13 January 2003

Abstract—Systematic studies were performed on the photo-Fries and the Fries reaction of aliphatic, aliphatic unsaturated, aromatic and aromatic unsaturated esters of 5,8-dihydro-1-naphthol. The Fries reaction of 5,8-dihydro-1-naphthyl acetate in various solvents is also reported. © 2003 Elsevier Science Ltd. All rights reserved.

Since the observation of photo-Fries rearrangement in 1960, various aspects of the reaction have been studied.^{1,2} Since then studies were initiated on the mechanistic aspects as well as the synthetic applications of photo-Fries reactions.^{3–6} Crouse et al. have studied the photo-Fries reaction of esters of 1-naphthol and reported various substituted naphthoquinones derived from the photo-Fries product and their results provide a convenient method for the construction of tricyclic analogues of adriamycinone.⁷ Recently, the photo-Fries reaction of tetrahydronaphthyl esters has also been reported.⁸ Herein, we report the studies on photo-Fries reaction of 5,8-dihydronaphthyl esters. The 5,8-di-hydro-1-naphthol system will be of great interest in view of the isolated double bond.

In continuation of our interest on the photo-Fries reaction,^{9,10} we have carried out the photochemical studies of various esters of 5,8-dihydro-1-naphthol (1). In the aliphatic series, the acetate (2a), crotonate (2b), and sorbate (2c) were prepared and their photochemistry studied. In the aromatic series, the benzoate (2d), 2-methylbenzoate (2e), 2-chloro-

benzoate (2f), 1-naphthoate (2g), and cinnamate (2h) were studied and the results of the studies are discussed here.

The irradiation of 5,8-dihydro-1-naphthyl esters (2a-h) was carried out for 24 h in an Applied Photophysics Reactor at 254 nm in methanol in a quartz vessel (Scheme 1).

The products were separated by column chromatography over silica gel. The *ortho* rearrangement product **3** and the *para* product **4** were obtained in about 3:1 ratio respectively, in addition to the 5,8-dihydronaphthol **1**. The products were characterized by spectral and analytical data. Likewise, the crotonate **2b**, on irradiation, furnished the respective *ortho*and *para*-rearrangement products **3b** and **4b**, in addition to 5,8-dihydronaphthol **1**. The sorbate **2c**, on irradiation also gave two products **3c** and **4c**.

Irradiation of the benzoate **2d** in methanol furnished a mixture, which was chromatographed to separate the two rearrangement products **3d** and **4d**: compound **3d** was found to be contaminated with the 5,8-dihydronaphthol **1**. The



Scheme 1. Reagents and conditions: (i) Li/NH₃, EtOH, 85%; (ii) RCOCl, benzene/pyridine, 80-90%; (iii) hv (254 nm)/MeOH.

Keywords: photo-Fries; dihydro-1-naphthyl esters, aromatic esters; photolysis; rearrangement.

^{*} Corresponding author. Tel.: +91-44-2351269x213; fax: +91-44-2352494; e-mail: vtrk28@yahoo.com, sriraghavan68@hotmail.com.

	Carle stars to 2	P	01 V:-14 - f 2	01 X:-1-1 - f 4	0/ V:-14 -£ 1
Entry	Substrate 2	ĸ	% ried of 3	% rield of 4	% field of I
1	а	CH ₃	40	12	25
2	b	CH=CH-CH ₃	32	12.5	30
3	с	(CH=CH)2-CH3	32	12	30
4	d	C ₆ H ₅	21.2 ^a	10	30
5	e	$2-CH_3C_6H_4$	35	_	30
6	f	$2-ClC_6H_4$	25	10	40
7	g	1-Naphthyl	30	20	30
8	ĥ	CH=CH-C ₆ H ₅	15	-	40

Table 1. Product distribution upon photolysis of 5,8-dihydro-1-naphthyl esters 2a-h

Irradiation studies were carried out at 254 nm for 24 h in an Applied Photophysics Photochemical Reactor using absolute methanol as solvent. ^a As benzoate.

mixture was benzoylated and the compound **3d** was isolated as its benzoyl derivative **5**. The 2-methylbenzoate **2e** on irradiation in benzene or methanol furnished only the *ortho* rearranged product **3e** along with the dihydronaphthol.



The 2-chlorobenzoate **2f** on photolysis furnished the two products **3f** and **4f** in addition to the dihydronaphthol. No cyclization product was isolated which can be expected in the light of other reports.¹¹ Irradiation of the ester **2f** in different solvents (benzene, ethanol, ethyl acetate) or at 300 nm gave the two rearranged products **3f** and **4f** only, with slight variation in their yields, but not any cyclized product. The 1-naphthoate **2g** gave the two products **3g** and **4g** as expected, on irradiation in methanol. The irradiation of the cinnamate **2h** in methanol furnished only the *ortho*-rearranged product **3h**, as observed earlier in similar systems¹⁰ (Table 1).

The esters $2\mathbf{a} - \mathbf{h}$ were characterized by IR and ¹H NMR and were directly used for the irradiation studies, since they were found to be slowly undergoing decomposition. The products were characterized by IR, ¹H NMR, Mass spectra and elemental analysis. The ¹H NMR of the 2-acyl compounds **3a** invariably showed the C₄-H around δ 6.5, as a doublet with coupling constant ranging between 7 and 9 Hz; the C₆-H and C₇-H appeared around δ 5.8 as a complex signal due to the coupling between the adjacent olefinic and methylene protons; the C₅ and C₈ methylene protons appeared around δ 3.3 as a complex signal due to geminal and vicinal coupling. The Ar-OH appeared around δ 12, which was exchanged with D₂O. The mass spectrum of products **3** and **4** invariably showed an intense molecular ion, in some cases as a base peak. The structure of compounds **3a**,¹² **3b**,¹³ **3g**¹⁴ and **4g**¹³ were also confirmed by X-ray crystallographic studies. Thus the photo-Fries reaction of esters of 5,8-dihydro-1-naphthol gave in general, the *ortho* and *para*-rearrangement products in moderate yields. Invariably, 5,8-dihydro-1-naphthol was obtained in the irradiations, similar to the formation of phenol in the photolysis of phenyl esters. Aromatization of the dihydro-1-naphthol ring did not occur in any of the cases in the irradiation studies.

The Fries reaction of 2-naphthyl alkanoate is known.¹⁵ The reaction of 2-naphthyl cinnamate in the presence of aluminium chloride has been studied by us earlier.¹⁰ The reaction of 1-naphthyl cinnamate in presence of aluminium chloride gave only 1-naphthol. Hence the Fries reaction of 5,8-dihydro-1-naphthyl acetate (**2a**) was next studied. The treatment of **2a** with anhydrous aluminium chloride in either nitrobenzene or carbon disulfide at room temperature furnished 4-acetyl-5,8-dihydro-1-naphthol (**4a**) in 30–35% yield, along with the 5,8-dihydronaphthol **1** (40%).

When benzene was used as a solvent, the ester 2a, in the presence of AlCl₃, gave two products **6** (35.7%) and **7** (10.7%), involving Fries rearrangement and arylation at double bond. The arylation has occurred similar to the earlier reports.^{16,17} The structures of compounds **6** and **7** were established by spectral and analytical data and the structure of compound **6** was further confirmed by X-ray crystallographic studies¹⁸ as its tosyl derivative **8** (Scheme 2).

In conclusion, irradiation of 5,8-dihydro-1-naphthyl esters in methanol at 254 nm furnished *ortho* and *para* rearranged products along with 5,8-dihydro-1-naphthol. The Fries



reaction of 5,8 dihydro-1-naphthyl acetate using benzene as the solvent, furnished the arylated Fries reaction product.

1. Experimental

Melting points were determined by using a Toshniwal melting point apparatus in an open capillary tube and are uncorrected. IR spectra were recorded in Nicolet Impact 400 FT-IR spectrophotometer. ¹H and ¹³C NMR spectra were recorded on Jeol 400 MHz, Bruker 300 MHz, Bruker ARX 200 MHz and Jeol FX 90Q spectrometers, using TMS as internal standard. Mass spectra were obtained from a Jeol-DX-303 spectrometer. Microanalyses were performed in a Perkin–Elmer 240B elemental analyzer. Chromatography was performed on silica gel (ACME, 100–200 mesh).

1.1. General procedure for the preparation of 5,8-dihydro-1-naphthlyl esters (2a-h)

To a solution of 5,8-dihydronaphthol **1** (3.65 g, 25 mmol), and pyridine (2.40 mL, 30 mmol) in dry benzene (50 mL), was added dropwise a solution of the corresponding acid chloride (30 mmol) in dry benzene (30 mL), under cooling at 10°C. The reaction mixture was stirred overnight and washed with dil. NaHCO₃ solution, dil. NaOH solution and then several times with water. The organic layer was dried over anhydrous magnesium sulphate, the solvent removed under vacuum and the crude ester purified either by column chromatography using hexane–ethyl acetate (9:1) as an eluant or by recrystallization.

1.1.1. 5,8-Dihydro-1-naphthyl acetate¹⁹ **(2a).** Following the general procedure, the reaction of 5,8-dihydro-1-naphthol (3.65 g, 25 mmol), with acetyl chloride (2.10 mL, 30 mmol) afforded the compound **2a** 4.2 g (90%) as a colourless liquid. ν_{max} (KBr) 3030, 2880, 2820, 1760, 1460, 1360, 1200 cm⁻¹; δ_{H} (90 MHz, CDCl₃) 6.9 (m, 3H, Ar-*H*), 5.7 (m, 2H, Ar-*H*₆, Ar-*H*₇), 3.3 (m, 2H, 5-C*H*₂), 3.1 (m, 2H, 8-C*H*₂), 2.4 (s, 3H, C*H*₃).

1.1.2. 5,8-Dihydro-1-naphthyl crotonate²⁰ (**2b**). The crotonyl chloride was prepared by refluxing crotonic acid (2.6 g, 30 mmol) and thionyl chloride (5 mL) in dry benzene for 3 h. The solvent was removed under reduced pressure along with the excess of thionyl chloride. The crude crotonyl chloride was used in the next step without further purification.

Following the general procedure, treatment of 5,8-dihydro-1-naphthol (3.65 g, 25 mmol) with crotonyl chloride and pyridine (2.40 mL, 30 mmol), afforded the title compound **2b** (4.7 g, 88%) as a colourless liquid. ν_{max} (KBr) 3015, 2880, 1735, 1650, 1570, 1460, 1310, 1290, 1220 cm⁻¹; $\delta_{\rm H}$ (90 MHz, CDCl₃) 6.9 (m, 3H, Ar-*H*), 6.7 (m, 2H), 5.8 (m, 2H, Ar-*H*₆, Ar-*H*₇), 3.5 (m, 2H, 5-C*H*₂), 3.2 (m, 2H, 8-C*H*₂), 2.0 (d, *J*=6 Hz, 3H, C*H*₃).

1.1.3. 5,8-Dihydro-1-naphthyl-2,4-hexadienoate (2c). 2,4-Hexadienoyl chloride (sorbyl chloride) was prepared by refluxing 2,4-hexadienoic acid (3.4 g, 30 mmol) and thionyl chloride (5 mL) in dry benzene for 3 h. The solvent

and the excess of thionyl chloride were removed under reduced pressure. The crude acid chloride was used for esterification of 5,8-dihydro-1-naphthol (3.65 g, 2.5 mmol) to afford the title compound **2c** (4.9 g 82%) as a white solid. Mp 58–60°C; $\nu_{\rm max}$ (KBr) 3000, 1720, 1630, 1450, 1320, 1210, 1110, 990 cm⁻¹; $\delta_{\rm H}$ (90 MHz, CDCl₃) 7.45 (1H, m, Ar-*H*), 7.15 (1H, m, Ar-*H*), 6.95 (2H, m, Ar-*H*, CH=C*H*), 6.26 (2H, m), 5.97 (1H, m), 5.87 (2H, m, *H*₆, *H*₇), 3.43 (2H, bs, 5-C*H*₂), 3.2 (2H, bs, 8-C*H*₂), 1.9 (3H, d, *J*=5.1 Hz, CH*Me*); *m/z* (EI) 240 (M+, 33), 144 (13), 128 (8), 117 (5), 116 (6), 115 (22), 96 (14), 95 (100), 67 (61), 66 (8), 65 (21), 51 (10).

1.1.4. 5,8-Dihydro-1-naphthyl benzoate (2d). The treatment of 5,8-dihydro-1-naphthol (3.4 g, 25 mmol), with benzoyl chloride (3.5 mL, 30 mmol) furnished the ester **2d** (4.6 g, 75%) as a colourless liquid. ν_{max} (KBr) 3020, 2880, 2820, 1730, 1450, 1060, 1020, 700 cm⁻¹; δ_{H} (90 MHz, CDCl₃) 8.1 (2H, m, Ar-*H*), 7.4 (4H, m, Ar-*H*), 6.9 (2H, m, Ar-*H*), 5.8 (2H, m, H₆, H₇), 3.3 (4H, m, 5 and 8-CH₂).

1.1.5. 5,8-Dihydro-1-naphthyl 2-methylbenzoate (2e). The 2-methylbenzoyl chloride was prepared by refluxing *o*-toluic acid (4 g, 30 mmol), and thionyl chloride (5 mL) in dry benzene for 3 h. The crude 2-methylbenzoyl chloride was treated with 5,8-dihydro-1-naphthol (3.65 g, 25 mmol) to afford the ester **2e** (5.6 g, 85%) as a colourless and viscous liquid. ν_{max} (KBr) 3000, 2870, 2825, 1728, 1610, 1450, 1320, 1125, 985,870 cm⁻¹; δ_{H} (90 MHz, CDCl₃) 7.85 (1H, m, Ar-*H*), 7.23 (6H, m, Ar-*H*), 5.8 (2H, m, *H*₆, H₇), 3.3 (4H, m, 5 and 8-CH₂), 2.4 (3H, s, CH₃).

1.1.6. 5,8-Dihydro-1-naphthyl 2-chlorobenzoate (2f). The 2-chlorobenzoyl chloride, obtained from *o*-chlorobenzoic acid (4.7 g, 30 mmol) and 5,8-dihydro-1-naphthol (3.65 g, 25 mmol) in the presence of pyridine (2.40 mL, 30 mmol) in dry benzene furnished the ester **2f** (5.8 g, 82%) as a viscous liquid. $\nu_{\rm max}$ (KBr) 3010, 2860, 1740, 1460, 1280, 1200, 970, 850 cm⁻¹; $\delta_{\rm H}$ (90 MHz, CDCl₃) 7.9 (1H, m, Ar-*H*), 7.20 (6H, m, Ar-*H*), 5.8 (2H, m, H_6 , H_7), 3.3 (4H, m, 5 and 8-C H_2).

1.1.7. 5,8-Dihydro-1-naphthyl 1-naphthoate (2g). The 1-naphthoyl chloride, prepared by refluxing 1-naphthoic acid (5.2 g, 30 mmol) and thionyl chloride (5 mL) in dry benzene for 3 h, on treatment with 5,8-dihydro-1-naphthol (3.65 g, 25 mmol) in the presence of pyridine (2.40 mL, 30 mmol) furnished the ester **2g** (4.3 g, 58%) as a crystalline solid. Mp 100–102°C; ν_{max} (KBr) 3010, 2890, 2860, 1730, 1450, 1300, 1210, 960, 850, 810 cm⁻¹; δ_{H} (90 MHz, CDCl₃) 8.90 (3H, m, Ar-H), 8.4 (1H, m, Ar-H), 7.8 (2H, m, Ar-H), 7.5 (3H, m, Ar-H), 5.8 (2H, m, H₆, H₇), 3.0 (4H, m, 5 and 8-CH₂); *m/z* (EI) 300 (M+, 11.4), 155 (100), 127 (38.2), 115 (6.32).

1.1.8. 5,8-Dihydro-1-naphthyl cinnamate (2h). The treatment of cinnamoyl chloride, from cinnamic acid (4.5 g, 30 mmol) and thionyl chloride (5 mL), with 5,8-dihydro-1-naphthol (3.65 g, 25 mmol) in the presence of pyridine (2.4 mL, 30 mmol) afforded the ester **2h** (5.4 g, 78%) as a colourless solid. Mp 122–124°C; ν_{max} (KBr) 1720, 1620, 1450, 1310, 1210, 1140, 1020, 870, 800, 750, 710 cm⁻¹; $\delta_{\rm H}$

(90 MHz, CDCl₃) 7.9 (1H, d, J=16 Hz, H_{β}), 7.65 (2H, m, Ar-*H*), 7.45 (3H, m, Ar-*H*), 7.0 (3H, m, Ar-*H*), 6.6 (1H, d, J=16 Hz, H_{α}), 5.9 (2H, m, H_6 , H_7), 3.45 (2H, m, 5-CH₂), 3.25 (2H, m, 8-CH₂); m/z (EI) 276 (M+, 3), 144 (2), 131 (100), 115 (4.3), 103 (27), 91 (2).

1.2. General procedure for the irradiation of 5,8-dihydro-1-naphthyl esters (2a-h)

A solution of 5,8-dihydro-1-naphthyl ester 3 (1.0 g) in methanol or benzene (150 mL) was purged with nitrogen for half an hour in a quartz vessel and irradiated at 254 nm using Applied Photo-physics photochemical reactor. The course of the reaction was followed by TLC. After the disappearance of the starting material, the solvent was removed under vacuum. The residue obtained was chromatographed over a column of silica gel and eluted with a mixture of hexane–ethyl acetate (9:1), to isolate the products.

1.3. Irradiation of 5,8-dihydro-1-naphthyl acetate (2a)

Photolysis of the ester 2a (1 g, 5.3 mmol) in methanol (150 mL) at 254 nm for 24 h gave the products 3a, 4a and 5,8-dihydro-1-naphthol 1 (25%).

1.3.1. 2-Acetyl-5,8-dihydro-1-naphthol (3a). Yield: 0.4 g (40%); mp 98–100°C; [Found: C, 75.95; H, 6.11. $C_{12}H_{12}O_2$ requires C, 76.57; H, 6.43%]; ν_{max} (KBr) 3020, 1620, 1490, 1430, 1410, 1360, 1320, 1270, 990, 840, 800 cm⁻¹; $\delta_{\rm H}$ (90 MHz, CDCl₃) 12.5 (1H, s, D₂O exchanged, Ar-OH), 7.46 (1H, d, *J*=9 Hz, Ar-*H*₃), 7.56 (1H, d, *J*=9 Hz, Ar-*H*₄), 5.80 (2H, m, *H*₆, *H*₇), 3.3 (4H, bs, 5 and 8-CH₂), 2.5 (3H, s, CH₃); *m*/*z* (EI) 188 (M+, 100), 187 (25), 186 (34), 173 (82), 171 (55), 155 (11), 145 (37), 127 (16), 115 (43), 105 (4), 91 (10), 86 (5).

1.3.2. 4-Acetyl-5,8-dihydro-1-naphthol (4a). Yield: 0.12 g (12%); mp 154–156°C; [Found: C, 76.20; H, 6.23. C₁₂H₁₂O₂ requires C, 76.57; H, 6.43%]; ν_{max} (KBr) 3035, 1640, 1495, 1400, 1420, 1350, 1250 cm⁻¹; δ_{H} (90 MHz, CDCl₃) 13.0 (1H, s, D₂O exchanged, Ar-OH), 7.40 (1H, d, J=9 Hz, Ar- H_3), 7.20 (1H, d, J=9 Hz, Ar- H_2), 5.75 (2H, m, H_6, H_7), 3.28 (4H, bs, 5 and 8-CH₂), 2.35 (3H, s, CH₃); m/z (EI) 188 (M+, 100), 173 (30), 145 (47), 127 (45), 115 (85), 91 (30), 77 (30), 63 (46), 57 (18).

1.4. Irradiation of 5,8-dihydro-1-naphthyl crotonate (2b)

Photolysis of the ester 2b (1.0 g, 4.67 mmol) in methanol at 254 nm for 24 h afforded the compounds 3b, 4b and 5,8-dihydro-1-naphthol (30%).

1.4.1. 2-Crotonyl-5,8-dihydro-1-naphthol (**3b**). Yield: 0.32 g (32%); mp 126–128°C; [Found: C, 78.07; H, 6.42. C₁₄H₁₄O₂ requires C, 78.48; H, 6.59%]; ν_{max} (KBr) 3400, 3050, 3000, 2900, 1630, 1590, 1420, 1350, 1300, 790 cm⁻¹; $\delta_{\rm H}$ (90 MHz, CDCl₃) 13.0 (1H, s, D₂O exchanged, Ar-OH), 7.50 (1H, d, *J*=9 Hz, Ar-H₃), 6.8–7.3 (2H, m, Ar-H), 6.6 (1H, d, *J*=9 Hz, H₄), 5.80 (2H, m, H₆ and H₇), 3.30 (4H, bs, H_5 and H_8), 2.0 (3H, d, *J*=6 Hz, CH₃); *m/z* (EI) 214 (M+, 18), 199 (100), 173 (25), 127 (7), 115 (21), 105 (3), 99 (6), 91 (7).

1.4.2. 4-Crotonyl-5,8-dihydro-1-naphthol (**4b**). Yield: 0.125 g (12.5%); mp 198–200°C; [Found: C, 78.01; H, 6.32. C₁₄H₁₄O₂ requires C, 78.48; H, 6.59%]; ν_{max} (KBr) 3420, 3048, 3000, 2950, 1650, 1450, 980 cm⁻¹; δ_{H} (90 MHz, CDCl₃) 13.2 (1H, s, D₂O exchanged, Ar-OH), 7.57 (1H, d, J=9 Hz, Ar-H₃), 7.0–7.20 (2H, m, H_{α} and H_{β}), 6.6 (1H, d, J=9 Hz, Ar-H₂), 5.75 (2H, m, H_6 and H_7), 3.50 (4H, bs, H_5 and H_8), 2.2 (3H, dd, J=6.1, 1.0 Hz, CH₃).

1.5. Irradiation of 5,8-dihydro-1-naphthyl 2,4-hexadienoate (2c)

Photolysis of **2c** (1 g, 4.16 mmol) in MeOH (150 mL) at 254 nm, for 24 h furnished the rearranged compounds **3c**, **4c** and 5,8-dihydro-1-naphthol (30%).

1.5.1. 2-(2,4-Hexadienoyl)-5,8-dihydro-1-naphthol (3c). Yield: 0.32 g (32%); mp 128–130°C; [Found: C, 79.35; H, 6.58. C₁₆H₁₆O₂ requires C, 79.97; H, 6.71%]; ν_{max} (KBr) 3450, 1620, 1610, 1600, 1575, 1400 cm⁻¹; $\delta_{\rm H}$ (90 MHz, CDCl₃) 12.8 (1H, s, D₂O exchanged, Ar-OH), 7.50 (1H, d, J=9 Hz, H_3), 6.80 (1H, m), 6.50 (1H, d, J=9 Hz, H_4), 6.20 (3H, m), 5.8 (2H, m, H_6 and H_7), 3.20 (4H, bs, H_5 and H_8), 1.8 (3H, d, J=6 Hz, CH_3); m/z (EI) 240 (M+, 0.3), 239 (5), 238 (49), 223 (18), 171 (25), 170 (100), 115 (76), 114 (48), 89 (16), 77 (5), 67 (29), 66 (24), 65 (43), 52 (14).

1.5.2. 4-(2,4-Hexadienoyl)-5,8-dihydro-1-naphthol (4c). Yield: 0.12 g (12%); mp 108–110°C; [Found: C, 79.49; H, 6.65. C₁₆H₁₆O₂ requires C, 79.97; H, 6.71%]; ν_{max} (KBr) 3420, 1635, 1612, 1600, 1560, 1500 cm⁻¹; δ_{H} (90 MHz, CDCl₃) 12.5 (1H, s, Ar-OH), 7.26 (1H, d, *J*=7.8 Hz, *H*₃), 6.69 (1H, d, *J*=8.2 Hz, *H*₂), 6.67 (1H, m), 6.49 (1H, m), 6.20 (2H, m), 5.88 (2H, bs, *H*₆ and *H*₇), 3.56 (2H, m, 5-CH₂), 3.32 (2H, m, 8-CH₂), 1.88 (3H, d, *J*=6 Hz, CH₃).

1.6. Irradiation of 5,8-dihydro-1-naphthyl benzoate (2d)

Photolysis of the ester **2d** (1 g, 4 mmol) in MeOH at 254 nm, for 24 h furnished 2-benzoyl-5,8-dihydro-1-naphthol (**3d**), 4-benzoyl-5,8-dihydro-1-naphthol (**4d**) and 5,8-dihydro-1-naphthol (30%). The separation of the *ortho* rearranged product was difficult by column chromatography, since its R_f value and that of 5,8-dihydro-1-naphthol were same. Hence the mixture of compounds **3d** and **1** was treated with benzoyl chloride (1.0 g) in the presence of pyridine (2 mL) using dry benzene as solvent at room temperature. After completion of the reaction, the solvent was removed and the residue was chromatographed over a silica gel column and eluted with hexane–ethyl acetate (9:1) to isolate the benzoyl derivative **5** as a colourless solid.

1.6.1. 2-Benzoyl-5,8-dihydro-1-naphthyl benzoate (5). Yield: 0.3 g (21%); mp 142–144°C; [Found: C, 81.25; H, 5.07. C₂₄H₁₈O₃ requires C, 81.34; H, 5.12%]; ν_{max} (KBr) 3020, 2890, 1720, 1630, 1600, 1590, 1420, 1350 cm⁻¹; $\delta_{\rm H}$ (90 MHz, CDCl₃) 7.7 (5H, m, Ar-*H*), 7.3 (7H, m, Ar-*H*), 5.9 (2H, m, *H*₆ and *H*₇), 3.4 (2H, m, 5-CH₂), 3.3 (2H, m, 8-CH₂); *m/z* (EI) 354 (M+, 4), 353 (7), 352 (35), 249 (49), 247 (30), 246 (7), 105 (100), 77 (90).

1.6.2. 4-Benzoyl-5,8-dihydro-1-naphthol (**4d**). Yield: 0.1 g (10%); mp 184–186°C; [Found: C, 81.60; H, 5.57.

C₁₇H₁₄O₂ requires C, 81.58; H, 5.63%]; ν_{max} (KBr) 3400, 3020, 2980, 1630, 1590, 1500, 1420, 970, 860 cm⁻¹; $\delta_{\rm H}$ (90 MHz, CDCl₃) 12.5 (1H, s, Ar-OH), 7.7 (4H, m, Ar-H), 7.0 (3H, m, Ar-H), 5.8 (2H, m, H₆ and H₇) 3.3 (4H, bs, 5 and 8-CH₂); *m*/z (EI) 250 (M+, 100), 171 (26), 145 (18), 127 (12), 115 (35), 105 (94), 91 (9), 77 (82).

1.7. Irradiation of 5,8-dihydro-1-naphthyl 2-methylbenzoate (2e)

Photolysis of the ester 2e (1 g, 3.8 mmol) in benzene (150 mL) at 254 nm, for 24 h furnished, the *ortho* rearranged product (3e) along with 5,8-dihydro-1-naphthol 1 (30%).

1.7.1. 2-(2-Methylbenzoyl-5,8-dihydro-1-naphthol (3e). Yield: 0.35 g (35%); mp 78–80°C; [Found: C, 81.20; H, 5.98. $C_{18}H_{16}O_2$ requires C, 81.79; H, 6.10%]; ν_{max} (KBr) 3400, 3025, 2890, 1630, 1590, 880, 870 cm⁻¹. δ_{H} (300 MHz, CDCl₃) 12.8 (1H, s, Ar-OH), 8.53 (1H, d, J=7.5 Hz, Ar- H_3), 7.65 (1H, m, Ar-H), 7.30 (2H, m, Ar-H), 7.10–7.75 (4H, m), 6.56 (1H, d, J=7.5 Hz, Ar- H_4), 5.90 (2H, m, H_6 and H_7), 3.40 (4H, bs, 5 and 8-C H_2), 2.31 (3H, s, Ar-C H_3); δ_c (75 MHz, CDCl₃) 204.0, 161.4, 143.9, 138.1, 135.2, 130.7, 129.6, 129.8, 127.3, 125.9, 118.9, 117.0, 113.4, 30.21, 23.6, 19.5; m/z (EI) 264 (M+, 65), 263 (12), 249 (69), 247 (65), 245 (21), 223 (15), 173 (13), 172 (14), 171 (35), 170 (53), 119 (51), 115 (48), 114 (18), 91 (100), 90 (13), 77 (10), 65 (57), 51 (22).

1.8. Irradiation of 5,8-dihydro-1-naphthyl 2-chlorobenzoate (2f)

Photolysis of ester 2f (1 g, 3.5 mmol) in MeOH at 254 nm, for 24 h furnished the products 3f, 4f and 5,8-dihydro-1-naphthol 1 (40%).

1.8.1. 2-Chlorobenzoyl-5,8-dihydro-1-naphthol (**3f**). Yield: 0.25 g (25%); mp 112–114°C; [Found: C, 72.08; H, 4.70. $C_{17}H_{13}ClO_2$ requires C, 71.71; H, 4.60%]; ν_{max} (KBr) 3420, 3020, 2980, 1620, 1600, 1520, 1480 cm⁻¹; δ_{H} (300 MHz, CDCl₃) 12.2 (1H, s, Ar-OH), 7.55 (4H, m, Ar-H), 7.0 (1H, d, *J*=9 Hz, *H*₃), 6.7 (1H, d, *J*=9 Hz, *H*₄), 5.9 (2H, m, *H*₆ and *H*₇), 3.4 (2H, bs, 8-CH₂), 3.2 (2H, bs, 5-CH₂); *m/z* (EI) 286 (M+2, 32), 284 (M+, 88), 283 (15), 282 (15), 250 (15), 249 (82), 247 (56), 173 (33), 171 (23), 145 (35), 141 (33), 139 (100), 115 (39), 111 (39), 91 (10).

1.8.2. 4-(2-Chlorobenzoyl)-5,8-dihydro-1-naphthol (4f). Yield: 0.1 g (10%); mp 210–212°C; [Found: C, 71.36; H, 4.52. $C_{17}H_{13}ClO_2$ requires C, 71.71; H, 4.60%]; ν_{max} (KBr) 3400, 3020, 2990, 1640, 1600, 1580, 1420 cm⁻¹; δ_{H} (300 MHz, DMSO- d_{6}) 10.6 (1H, s, Ar-OH), 7.5 (4H, m, Ar-H), 7.05 (1H, d, J=9.0 Hz, H₃), 6.7 (1H, d, J=9.0 Hz, H₂), 5.9 (2H, m, H₆ and H₇), 3.25 (2H, m, 8-CH₂), 3.2 (2H, m, 5-CH₂); m/z (EI) 286 (M+2, 16), 284 (M+, 24), 283 (24), 282 (2), 251 (53), 249 (48), 231 (25), 175 (6), 145 (32), 141 (33) 139 (100), 127 (14), 115 (25), 111 (32), 91 (7.7).

1.9. Irradiation of 5,8-dihydro-1-naphthyl α -naphthoate (2g)

Photolysis of 2g (0.64 g, 2.1 mmol) in benzene (150 mL) at

254 nm, for 24 h furnished the compounds **3g**, **4g** and 5,8-dihydro-1-naphthol **1** (30%).

1.9.1. 2-(1-Naphthoyl)-5,8-dihydro-1-naphthol (**3g**). Yield: 0.19 g (30%); mp 130–132°C; [Found: C, 84.42; H, 5.30. $C_{21}H_{16}O_2$ requires C, 83.98; H, 5.37%]; ν_{max} (KBr) 3420, 3200, 2990, 1630, 1620, 1590 cm⁻¹; δ_{H} (300 MHz, CDCl₃) 12.87 (1H, s, D₂O exchanged, Ar-OH), 7.98 (3H, m, Ar-H), 7.51 (4H, m, Ar-H), 7.08 (1H, bs, Ar-H), 6.49 (1H, bs, Ar-H₄), 5.9 (1H, bs, H₆), 5.85 (1H, bs, H₇), 3.40 (4H, bs, 5 and 8CH₂); δ_{c} (75 MHz, CDCl₃) 203.3, 161.5, 144.1, 136.0, 133.5, 130.8, 130.6, 130.4, 128.4, 127.1, 126.5, 126.1, 125.5, 125.4, 124.5, 124.4, 123.8, 122.9, 118.9, 117.6, 30.3, 23.6; *m/z* (EI) 300 (M+, 100), 283 (5), 281 (9), 173 (18), 172 (50), 171 (36), 155 (35), 144 (15), 128 (27), 127 (49), 77 (11).

1.9.2. 4-(1-Naphthoyl)-5,8-dihydro-1-naphthol (4g). Yield: 0.125 g (20%); mp 206–208°C; [Found: C, 84.08; H, 5.40. C₂₁H₁₆O₂ requires C, 83.98; H, 5.37%]; ν_{max} (KBr) 3400, 3000, 2920, 2850, 1620, 1420, 1330 cm⁻¹; $\delta_{\rm H}$ (300 MHz, CDCl₃) 10.40 (1H, s, Ar-OH), 8.02 (3H, m, Ar-H), 7.50 (4H, m, Ar-H), 7.06 (1H, m, Ar-H), 6.66 (1H, m, Ar-H), 5.9 (2H, m, H₆ and H₇), 3.72 (2H, m, 5-CH₂), 3.25 (2H, m, 8-CH₂); $\delta_{\rm c}$ (75 MHz, DMSO-d₆) 158.49, 138.4, 137.0, 133.2, 132.5, 130.6, 130.1, 128.4, 128.2, 128.1, 127.3, 127.1, 126.3, 125.32, 124.8, 124.2, 123.0, 122.2, 111.1, 28.3, 23.9; *m/z* (EI) 300 (M+, 100), 299 (13), 283 (13), 281 (13), 172 (15), 171 (91), 156 (11), 155 (77), 145 (16), 144 (10), 143 (13), 128 (20), 127 (82), 126 (37), 77 (17), 51 (16).

1.10. Irradiation of 5,8-dihydro-1-naphthyl cinnamate (2h)

Photolysis of the ester **2h** (1 g, 3.62 mmol) in MeOH at 254 nm, for 24 h furnished only the *ortho* rearranged product **3h** in low yield along with 5,8-dihydro-1-naphthol **1** (40%).

1.10.1. 2-Cinnamoyl-5,8-dihydro-1-naphthol (3h). Yield: 0.15 g (15%); mp 116–118°C; [Found: C, 82.08; H, 5.50. C₁₉H₁₆O₂ requires C, 82.58; H, 5.83 %]; ν_{max} (KBr) 3040, 1630, 1570, 1580, 1420, 1350, 1280, 1120, 760 cm⁻¹; $\delta_{\rm H}$ (90 MHz, CDCl₃) 13.0 (1H, s, D₂O exchanged, Ar-OH), 7.5 (5H, m, Ar-H), 7.3 (3H, m, Ar-H), 6.6 (1H, d, *J*=9 Hz, *H*₄), 5.80 (2H, m, *H*₆ and *H*₇), 3.3 (4H, bs, 5 and 8-CH₂); *m*/*z* (EI) 276 (M+, 63), 275 (33), 257 (8), 199 (36), 197 (15), 173 (14), 172 (34), 171 (51), 170 (100), 144 (22), 131 (43), 115 (37), 103 (36), 91 (16).

1.11. Reaction of 5,8-dihydro-1-naphthyl acetate (2a) with AlCl₃ in benzene

Anhydrous aluminium chloride (4.2 g, 32 mmol) was added in portion to the solution of the ester 2a (1 g, 5.3 mmol) in dry benzene (30 mL) at room temperature. Then the reaction mixture was refluxed on a water bath. After the disappearance of the starting material monitered by tlc, the reaction mixture was cooled and ice water (50 mL) added. The organic layer was separated and washed several times with dil. HCl and water and dried over magnesium sulphate. After distilling off the solvent, the residue was chromatographed over a silica gel column and eluted with 1796

hexane-ethyl acetate (4:1), to obtain the compounds **6** and **7**.

1.11.1. 2-Acetyl-6-phenyl-5,6,7,8-tetrahydro-1-naphthol (6). Yield: 0.5 g (35.7%); mp 118–120°C; [Found: C, 80.80; H, 6.73. $C_{18}H_{18}O_2$ requires C, 81.17; H, 6.81%]; ν_{max} (KBr) 3400, 3000, 2910, 1620, 1500, 1390, 1320, 800 cm⁻¹; $\delta_{\rm H}$ (300 MHz, CDCl₃) 12.90 (1H, s, Ar-OH), 7.50 (1H, d, *J*=8.3 Hz, H₃), 7.32 (5H, m, Ar-H), 6.65 (1H, d, *J*=8.3 Hz, H₄), 3.0 (4H, m), 2.9 (1H, m), 2.8 (3H, s, CH₃), 2.2 (1H, m), 1.90 (1H, m); δ_c (75 MHz, DMSO- d_6) 203.9, 160.6, 146.2, 128.6, 127.3, 126.8, 126.3, 119.4, 39.8, 38.3, 29.6, 26.4, and 23.1; *m/z* (EI) 266 (M+, 48), 251 (16), 175 (16), 162 (100), 147 (18), 115 (10), 104 (5), 91 (26).

1.11.2. 4-Acetyl-6-phenyl-5, 6, 7, 8-tetrahydro-1-naphthol (7). Yield: 0.15 g (10.7%); mp 218–220°C; [Found: C, 81.76; H, 7.01. $C_{18}H_{18}O_2$ requires C, 81.17; H, 6.81%]; ν_{max} (KBr) 3100, 2920, 1625, 1540, 1250, 1010, 810, 700 cm⁻¹; $\delta_{\rm H}$ (300 MHz, CDCl₃) 10.20 (1H, s, Ar-OH), 7.65 (1H, d, J=8.4 Hz, H_3), 7.3 (5H, m, Ar-H), 6.75 (1H, d, J=8.4 Hz, H_2), 3.25 (2H, m), 2.9 (2H, m), 2.70 (1H, m), 2.4 (3H, s, CH₃), 2.0 (1H, m), 1.8 (1H, m); m/z (EI) 266 (M+, 63), 251 (37), 190 (27), 175 (100), 162 (77), 147 (20), 133 (80, 115 (13), 91 (68).

1.11.3. 2-Acetyl-6-phenyl-5,6,7,8-tetrahydronaphthalen-1-yl 4-toluene sulphonate (8). To a suspension of NaH (0.038 g, 0.8 mmol) in THF (20 mL) was added dropwise a solution of the naphthol 6 (0.1 g, 0.37 mmol) in THF (20 mL), followed by the addition of *p*-toluenesulfonyl chloride (0.15 g, 0.75 mmol) in THF (5 mL) under nitrogen atmosphere at room temperature. After completion of the reaction, the reaction mixture was quenched with dil. NH₄Cl solution. The organic layer was separated, dried and chromatographed to obtain compound 8 and it was recrystallized from hexane–ethyl acetate mixture (1:1).

Yield: 0.14 g (90%); mp 98–100°C; ν_{max} (KBr) 3020, 1695, 1610, 1380, 1200, 1020, 900, 800, 790 cm⁻¹; $\delta_{\rm H}$ (300 MHz, CDCl₃) 7.75 (2H, d, *J*=8 Hz, Ar-*H*), 7.20–7.40 (8H, m, Ar-*H*), 7.10 (1H, d, *J*=8 Hz, *H*₄), 3.1 (1H, m), 2.46 (6H, s, 2CH₃), 2.91 (3H, m), 2.62 (1H, m), 2.1 (1H, m), 1.74 (1H, m); $\delta_{\rm c}$ (75 MHz, CDCl₃) 198.9, 145.5, 145.3, 144.1, 142.9, 132.9, 132.4, 132.0, 129.7, 128.3, 128.2, 127.4, 126.5, 126.2, 126.2, 39.3, 29.4, 29.2, 24.1, 21.5; $\delta_{\rm c}$ (75 MHz, DEPT-135, CDCl₃) 129.7, 128.3, 128.2, 126.4, 39.3, 37.4 (\downarrow), 29.2 (\downarrow), 24.1(\downarrow), 21.5.

Acknowledgements

The authors thank the UGC, New Delhi for Special Assistance Programme; RSIC, I.I.T., Chennai and SPIC Science Foundation, Chennai, for NMR facilities.

References

- Bellus, D. Advances in Photochemistry; Pitts, J. N., Hammond, G. S., Noyes, W. A. Jr., Eds.; Wiley-Interscience: New York, 1971; Vol. 8, p 109.
- Stenberg, V. I. Organic Photochemistry; Chapman, O. L., Ed.; Marcel Inc: NewYork, 1967; Vol. 1, p 127.
- 3. Coyle, J. D. Chem. Rev. 1978, 78, 97.
- 4. Szmant, H. Organic Building Blocks of the Chemical Industry; Wiley: New York, 1989; p 504.
- 5. Pitchumani, K.; Warrier, M.; Ramamurthy, V. J. Am. Chem. Soc. **1996**, 118, 9428, and references cited therein.
- Jimenez, M. C.; Leal, P.; Miranda, M. A.; Tormus, R. J. J. Chem. Soc., Chem. Commun. 1995, 2009.
- Crouse, D. J.; Hurlbut, S. L.; Wheeler, D. M. S. J. Org. Chem. 1981, 46, 374.
- 8. Gu, W.; Weiss, R. G. J. Org. Chem. 2001, 66, 1775.
- 9. Ramakrishnan, V. T.; Kagan, J. J. Org. Chem. 1970, 35, 2901.
- Jayachandran, T.; Manimaran, T.; Ramakrishnan, V. T. Proc. Indian Acad. Sci. 1986, 97, 41.
- 11. Grimshaw, J.; Prasanna de silva, A. J. Chem. Soc., Chem. Commun. 1980, 302.
- Chinnakali, K.; Fun, H. K.; Sriraghavan, K.; Ramakrishnan, V. T. Acta Crystallogr. 1998, C54, 370.
- Chinnakali, K.; Fun, H. K.; Sriraghavan, K.; Ramakrishnan, V. T. Acta Crystallogr. 1998, C54, 1496.
- Chinnakali, K.; Fun, H. K.; Sriraghavan, K.; Ramakrishnan, V. T. Acta Crystallogr. 1998, C54, 783.
- 15. Cui, C.; Wang, X.; Weiss, R. G. J. Org. Chem. 1996, 61, 1962.
- Corson, B. B.; Ipatieff, V. N. Organic Synthesis; Wiley: New York, 1955; Coll. Vol. II. p 151.
- 17. Wang, T. C.; Chen, Y. L.; Lee, K. H.; Tzeng, C. C. Synthesis 1997, 870.
- Chinnakali, K.; Fun, H. K.; Sriraghavan, K.; Ramakrishnan, V. T. Acta Crystallogr. 1999, C55, 944.
- Hauck, F. P.; Cimarusti, C. M.; Narayanan, V. L. US Appl. 01 Dec, 1971, 203, 865, 10 pp; *Chem Abstr.* 1974, 80, 95915r.
- Woodcock, D.; Byrde, R. J. W. Mededel. Landhouwhogeschool Opzoekingssta. Gent. 1963, 28, 568–573, Chem Abstr. 1966, 64, 14888b.